

2,6-Anhydro-1-deoxy-3,4-O-isopropylidene- β -D-ribo-hex-2-ulo-furanose

Louise A. Curran,^a Sarah F. Jenkinson,^{a*} Nigel A. Jones,^a David J. Watkin^b and George W. J. Fleet^a

^aDepartment of Organic Chemistry, Chemistry Research Laboratory, University of Oxford, 12 Mansfield Road, Oxford OX1 3TA, England, and ^bDepartment of Chemical Crystallography, Chemistry Research Laboratory, University of Oxford, 12 Mansfield Road, Oxford OX1 3TA, England
Correspondence e-mail: sarah.jenkinson@chem.ox.ac.uk

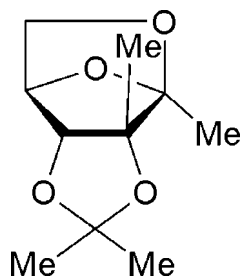
Received 27 June 2007; accepted 27 June 2007

Key indicators: single-crystal X-ray study; $T = 150$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.031; wR factor = 0.074; data-to-parameter ratio = 10.8.

The relative configuration of the title compound, $\text{C}_{10}\text{H}_{16}\text{O}_4$, was firmly established by X-ray crystallographic analysis. The absolute configuration was determined by the use of 2-*C*-methyl-D-ribo-1,4-lactone as the starting material.

Related literature

For related literature see: Curran *et al.* (2007); Jones *et al.* (2007); Hotchkiss *et al.* (2006).



Experimental

Crystal data

$\text{C}_{10}\text{H}_{16}\text{O}_4$	$V = 1041.05$ (6) Å ³
$M_r = 200.23$	$Z = 4$
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
$a = 6.6307$ (2) Å	$\mu = 0.10$ mm ⁻¹
$b = 11.1029$ (4) Å	$T = 150$ K
$c = 14.1409$ (5) Å	$0.60 \times 0.50 \times 0.10$ mm

Data collection

Nonius KappaCCD diffractometer	6557 measured reflections
Absorption correction: multi-scan	1370 independent reflections
<i>DENZO/SCALEPACK</i>	1160 reflections with $I > 2\sigma(I)$
(Otwinowski & Minor, 1997)	$R_{\text{int}} = 0.045$
$T_{\text{min}} = 0.95$, $T_{\text{max}} = 1.00$	

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.031$	127 parameters
$wR(F^2) = 0.074$	H-atom parameters constrained
$S = 0.93$	$\Delta\rho_{\text{max}} = 0.24$ e Å ⁻³
1370 reflections	$\Delta\rho_{\text{min}} = -0.19$ e Å ⁻³

Data collection: *COLLECT* (Nonius, 2001); cell refinement: *DENZO/SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO/SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2449).

References

- Altomare, A., Cascarano, G., Giacovazzo, G., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. & Watkin, D. J. (2003). *J. Appl. Cryst.* **36**, 1487.
- Curran, L. A., Jenkinson, S. F., Jones, N. A., Watkin, D. J. & Fleet, G. W. J. (2007). *Acta Cryst.* **E63**, o3387.
- Hotchkiss, D. J., Jenkinson, S. F., Storer, R., Heinz, T. & Fleet, G. W. J. (2006). *Tetrahedron Lett.* **47**, 315–318.
- Jones, N. A., Curran, L. A., Wormald, M. R., Dwek, R. A. & Fleet, G. W. J. (2007). In preparation.
- Nonius (2001). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography, Part A*, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). *CAMERON*. Chemical Crystallography Laboratory, University of Oxford, England.

supplementary materials

Acta Cryst. (2007). E63, o3388 [doi:10.1107/S1600536807031546]

2,6-Anhydro-1-deoxy-3,4-*O*-isopropylidene-3-*C*-methyl- β -D-ribo-hex-2-ulofuranose

L. A. Curran, S. F. Jenkinson, N. A. Jones, D. J. Watkin and G. W. J. Fleet

Comment

For background information see the previous paper (Curran *et al.*, 2007). The crystal structure of the title compound (Fig. 1) exists as alternating layers of molecules running parallel to the *ab*-face (Fig.2). There is no hydrogen bonding.

Experimental

Treatment of 1-deoxy-3-*C*-methyl-D-psicose **2** (Jones *et al.* in preparation) derived from 2-*C*-methyl-D-ribo-1,4-lactone **1** (Hotchkiss *et al.*, 2006), with sodium cyanide, gave a mixture of 2,6-anhydro derivative **3** and lactone **4** (Curran *et al.* 2007) (Fig. 3). X-ray analysis firmly established the structure of the title compound as the 2,6-anhydro furanose **3**. m.p. 321–326 K; $[\alpha]_D^{22}$ -47.0 (*c*, 1.0 in acetone).

Refinement

In the absence of significant anomalous scattering, Friedel pairs were merged and the absolute configuration was assigned from the starting material.

The H atoms were all located in a difference map, but those attached to carbon atoms were repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their geometry (C—H in the range 0.93–0.98, O—H = 0.82 Å) and $U_{\text{iso}}(\text{H})$ (in the range 1.2–1.5 times U_{eq} of the parent atom), after which the positions were refined with riding constraints.

Figures

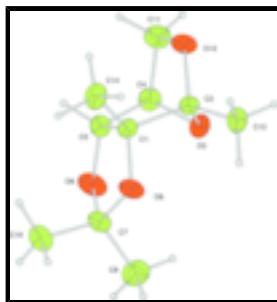


Fig. 1. The title compound with displacement ellipsoids drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radius.

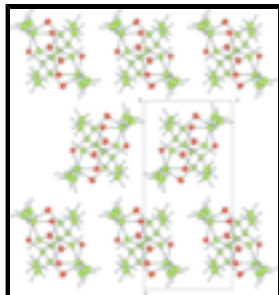


Fig. 2. Packing of the title compound projected along the *b*-axis.



Fig. 3. The reaction scheme.

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Crystal data

$C_{10}H_{16}O_4$	$F_{000} = 432$
$M_r = 200.23$	$D_x = 1.277 \text{ Mg m}^{-3}$
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
Hall symbol: P 2ac 2ab	$\lambda = 0.71073 \text{ \AA}$
$a = 6.6307 (2) \text{ \AA}$	Cell parameters from 1321 reflections
$b = 11.1029 (4) \text{ \AA}$	$\theta = 5\text{--}27^\circ$
$c = 14.1409 (5) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$V = 1041.05 (6) \text{ \AA}^3$	$T = 150 \text{ K}$
$Z = 4$	Plate, colourless
	$0.60 \times 0.50 \times 0.10 \text{ mm}$

Data collection

Nonius KappaCCD diffractometer	1160 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.045$
$T = 150 \text{ K}$	$\theta_{\text{max}} = 27.5^\circ$
ω scans	$\theta_{\text{min}} = 5.3^\circ$
Absorption correction: multi-scan	$h = -8 \rightarrow 8$
DENZO/SCALEPACK (Otwinowski & Minor, 1997)	$k = -14 \rightarrow 14$
$T_{\text{min}} = 0.95, T_{\text{max}} = 1.00$	$l = -18 \rightarrow 18$
6557 measured reflections	
1370 independent reflections	

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.031$	$w = 1/[\sigma^2(F^2) + (0.03P)^2 + 0.18P]$,
$wR(F^2) = 0.074$	where $P = [\max(F_o^2, 0) + 2F_c^2]/3$
	$(\Delta/\sigma)_{\text{max}} = 0.0002$

$S = 0.93$

$\Delta\rho_{\max} = 0.24 \text{ e } \text{\AA}^{-3}$

1370 reflections

$\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$

127 parameters

Extinction correction: None

Primary atom site location: structure-invariant direct methods

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.3568 (2)	0.21011 (14)	0.20100 (10)	0.0212
C2	0.4422 (2)	0.31928 (14)	0.14619 (10)	0.0217
O3	0.56240 (17)	0.37828 (10)	0.21534 (8)	0.0263
C4	0.6883 (2)	0.27622 (15)	0.23612 (11)	0.0272
C5	0.5351 (2)	0.18346 (15)	0.26887 (10)	0.0245
O6	0.46475 (18)	0.21350 (13)	0.36068 (7)	0.0352
C7	0.2509 (3)	0.22921 (16)	0.35888 (11)	0.0270
O8	0.19903 (16)	0.25108 (11)	0.26189 (7)	0.0283
C9	0.1985 (3)	0.34073 (17)	0.41531 (13)	0.0403
C10	0.1483 (3)	0.11679 (18)	0.39421 (13)	0.0396
C11	0.7600 (3)	0.24199 (19)	0.13709 (12)	0.0353
O12	0.58629 (17)	0.27261 (11)	0.07985 (8)	0.0317
C13	0.2974 (3)	0.40276 (16)	0.09881 (12)	0.0305
C14	0.2857 (3)	0.10662 (15)	0.13904 (12)	0.0308
H41	0.8009	0.2937	0.2816	0.0355*
H51	0.5810	0.0978	0.2636	0.0310*
H91	0.0522	0.3542	0.4084	0.0621*
H92	0.2310	0.3255	0.4825	0.0614*
H93	0.2804	0.4099	0.3910	0.0617*
H101	0.0017	0.1299	0.3883	0.0594*
H102	0.1804	0.1039	0.4613	0.0595*
H103	0.1903	0.0467	0.3550	0.0594*
H111	0.7881	0.1541	0.1327	0.0480*
H112	0.8841	0.2875	0.1191	0.0473*
H131	0.3743	0.4706	0.0698	0.0479*
H132	0.1972	0.4328	0.1459	0.0487*
H133	0.2281	0.3599	0.0465	0.0489*
H141	0.2242	0.0438	0.1796	0.0485*
H142	0.4037	0.0743	0.1055	0.0487*
H143	0.1825	0.1383	0.0923	0.0481*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0180 (7)	0.0224 (8)	0.0231 (7)	0.0020 (7)	-0.0018 (6)	-0.0006 (7)
C2	0.0217 (7)	0.0228 (8)	0.0207 (7)	0.0003 (7)	-0.0004 (7)	-0.0020 (6)
O3	0.0255 (6)	0.0219 (5)	0.0316 (6)	-0.0021 (5)	-0.0053 (5)	-0.0018 (5)
C4	0.0185 (7)	0.0304 (9)	0.0326 (8)	0.0012 (8)	-0.0031 (7)	0.0014 (8)
C5	0.0219 (8)	0.0267 (8)	0.0248 (7)	0.0014 (8)	-0.0033 (7)	0.0028 (7)

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O6	0.0274 (6)	0.0560 (8)	0.0221 (5)	0.0010 (7)	-0.0034 (5)	0.0044 (6)
C7	0.0257 (8)	0.0348 (10)	0.0204 (7)	-0.0004 (8)	-0.0008 (6)	0.0021 (7)
O8	0.0205 (5)	0.0431 (7)	0.0214 (5)	0.0029 (6)	0.0011 (5)	0.0025 (5)
C9	0.0467 (12)	0.0394 (10)	0.0347 (9)	0.0016 (10)	-0.0041 (9)	-0.0070 (8)
C10	0.0441 (11)	0.0375 (10)	0.0372 (9)	-0.0035 (10)	0.0063 (9)	0.0071 (9)
C11	0.0221 (8)	0.0443 (11)	0.0394 (10)	0.0043 (9)	0.0037 (8)	0.0026 (9)
O12	0.0273 (6)	0.0415 (7)	0.0262 (5)	0.0031 (6)	0.0057 (5)	0.0001 (6)
C13	0.0350 (10)	0.0261 (8)	0.0304 (8)	0.0022 (9)	-0.0060 (8)	0.0045 (7)
C14	0.0323 (9)	0.0254 (8)	0.0348 (9)	-0.0029 (8)	-0.0085 (8)	-0.0010 (8)

Geometric parameters (Å, °)

C1—C2	1.546 (2)	C7—C10	1.507 (3)
C1—C5	1.551 (2)	C9—H91	0.986
C1—O8	1.4293 (18)	C9—H92	0.988
C1—C14	1.520 (2)	C9—H93	1.001
C2—O3	1.4214 (18)	C10—H101	0.986
C2—O12	1.4357 (18)	C10—H102	0.983
C2—C13	1.493 (2)	C10—H103	0.995
O3—C4	1.438 (2)	C11—O12	1.448 (2)
C4—C5	1.519 (2)	C11—H111	0.995
C4—C11	1.527 (2)	C11—H112	0.998
C4—H41	1.004	C13—H131	0.998
C5—O6	1.4193 (19)	C13—H132	0.999
C5—H51	1.001	C13—H133	0.993
O6—C7	1.429 (2)	C14—H141	0.991
C7—O8	1.4348 (18)	C14—H142	0.983
C7—C9	1.514 (2)	C14—H143	1.014
C2—C1—C5	100.40 (12)	C7—O8—C1	110.27 (12)
C2—C1—O8	108.70 (12)	C7—C9—H91	107.3
C5—C1—O8	104.24 (11)	C7—C9—H92	108.4
C2—C1—C14	114.66 (13)	H91—C9—H92	109.6
C5—C1—C14	116.67 (13)	C7—C9—H93	108.8
O8—C1—C14	111.16 (13)	H91—C9—H93	112.5
C1—C2—O3	102.81 (11)	H92—C9—H93	110.0
C1—C2—O12	106.76 (12)	C7—C10—H101	107.1
O3—C2—O12	104.05 (12)	C7—C10—H102	110.0
C1—C2—C13	118.43 (14)	H101—C10—H102	108.4
O3—C2—C13	112.53 (13)	C7—C10—H103	109.6
O12—C2—C13	111.02 (13)	H101—C10—H103	110.2
C2—O3—C4	95.90 (11)	H102—C10—H103	111.4
O3—C4—C5	102.05 (13)	C4—C11—O12	101.92 (12)
O3—C4—C11	100.93 (13)	C4—C11—H111	111.1
C5—C4—C11	108.61 (15)	O12—C11—H111	110.1
O3—C4—H41	114.2	C4—C11—H112	111.4
C5—C4—H41	115.7	O12—C11—H112	113.2
C11—C4—H41	113.8	H111—C11—H112	109.0
C1—C5—C4	101.05 (12)	C11—O12—C2	104.40 (11)
C1—C5—O6	105.70 (12)	C2—C13—H131	108.9

C4—C5—O6	109.83 (14)	C2—C13—H132	109.6
C1—C5—H51	111.5	H131—C13—H132	111.3
C4—C5—H51	114.7	C2—C13—H133	109.5
O6—C5—H51	113.0	H131—C13—H133	107.0
C5—O6—C7	109.79 (12)	H132—C13—H133	110.5
O6—C7—O8	106.00 (12)	C1—C14—H141	109.1
O6—C7—C9	108.58 (16)	C1—C14—H142	107.9
O8—C7—C9	108.08 (14)	H141—C14—H142	110.5
O6—C7—C10	109.95 (16)	C1—C14—H143	108.8
O8—C7—C10	110.41 (14)	H141—C14—H143	110.0
C9—C7—C10	113.52 (15)	H142—C14—H143	110.5

Fig. 1

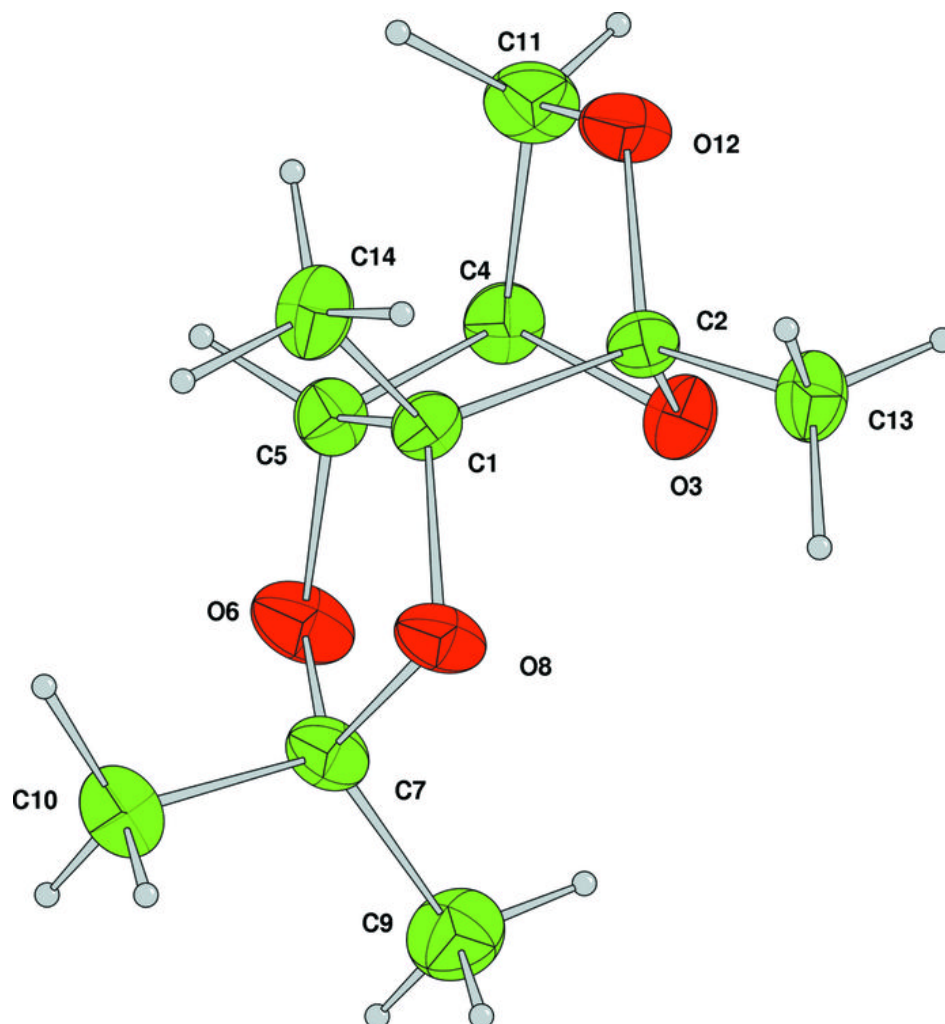


Fig. 2

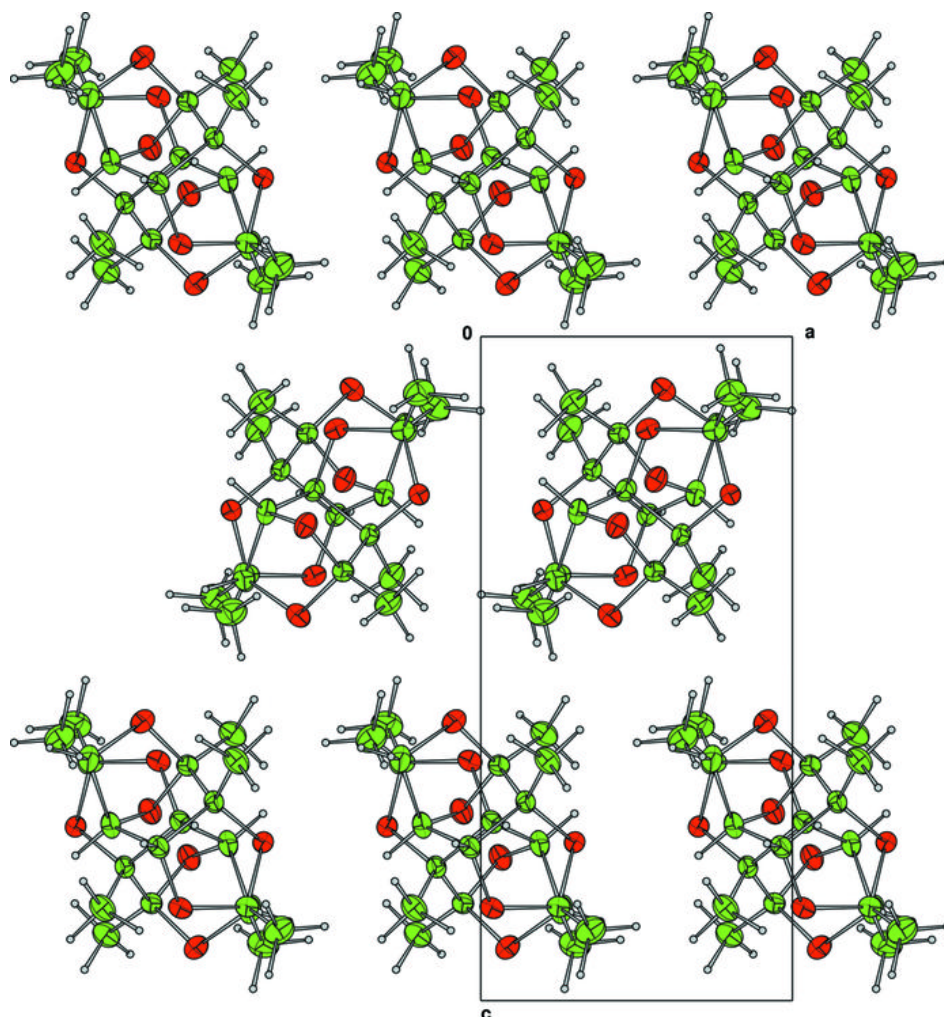


Fig. 3

